## Year 9: ASK Yourself!

## Subject: Physics Unit: 4 – Atomic Structure

	Structure					
	Launching 1-2	Developing 3-4	Progressing 5-6	Mastering		
	1-2	5-4	5-6	7-9		
<b>S</b> kills						
<b>N</b> kills						
	To be able to state that the number of protons in an element is the atomic number and the total number of protons and neutrons is the mass number. To be able to recognise the symbols used in a nuclear equation. To be able to define nuclear fission.	To be able to <u>u</u> nderstand that atoms of an element all have the same number of protons but can have different numbers of neutrons, giving different isotopes. To be able to write nuclear equations involving alpha and beta decay. To be able to describe nuclear fission using a nuclear equation.	To be able to use nuclear notation to show subatomic particles in an isotope. To be able to write balanced nuclear equations for different types of nuclear reaction. To be able to describe, using a nuclear equation, how an uncontrolled nuclear fission chain reaction can result in a huge release of energy.	To be able to calculate the half- life of a radioisotope. To be able to solve for unknown particles or elements in nuclear equations. To be able to describe methods used in nuclear power plants to control nuclear fission.		
<b>K</b> nowledge						
	To be able to recognise that some isotopes called radioisotopes are unstable and decay. To be able to list the three types of ionising radiation resulting from nuclear activity. To be able to explain the meaning of background radiation.	To be able to list some uses of radioisotopes in medicine. To be able to describe the structure of each type of ionising radiation. To be able to list different sources of background radiation. To be able to list the hazards of radioactive contamination.	To be able to describe how specific radioisotopes are used medicine. To be able to explain the properties of each type of radiation. To be able to explain why background radiation varies in different areas and in different times. To be able to compare and	To be able to explain the meaning of half- life of a radioisotope and selection based on half-life. To be able to choose between types of radiation for different applications. To be able to explain how radiation dose is measured and monitored.		

		To be able to define radioactive.		contrast irradiation and contamination.	To be able to explain the potential benefits of nuclear fusion
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