	g, Structure an Launching 1-2	Developing 3-4	Progressing 5-6	Mastering 7-9
S kills				
	To be able to recognise substances as small molecules, polymers or giant structures from diagrams showing their bonding. To be able to calculate areas of triangles and rectangles. To be able to use ratios, fractions and percentages. To be able to use SI units.	To be able to visualise and represent 2D and 3D forms including 2D representations of 3D objects. To be able to calculate surface area and volumes of cubes. To be able to recognise and use expressions in standard form. To be able to draw and interpret graphs.	To be able to make orders of magnitude calculations and use prefixes and powers of ten for orders of magnitude. To be able to interconvert units. To be able to use an appropriate number of significant figures in calculations.	To be able to explain applications of science and make decisions based on evidence and arguments. To use a variety of models appropriately such as representational, spatial, descriptive, and mathematical to solve problems, mak predictions and develop explanations
K nowledge				
	To know the three types of chemical bonds. Need to work on how the three types of bonding are different to each other. To know the three states of matter and how they transition between states.	To be able to clearly describe the difference between ionic, covalent and metallic bonding. To be able to describe properties of giant ionic, metallic and covalent structures. To be able to describe properties of diamond, graphite and polymers. To know some examples of nanoparticles.	To be able to represent ionic and covalent bonding in dot and cross diagrams. To be able to explain properties of giant ionic, metallic and covalent structures. To be able to explain the structure and properties of diamond, graphite and polymers. To be able to give examples of the use of nanoparticles. To have an awareness of intermolecular	To be able to work out the empirical formula of an ionic compound. To be able to explain how the bonding in diamond graphite and polymers affect their properties. To be able to describe bonding and properties of nanoparticles. To be able to explain how intermolecular forces affect physical properties